

$^1\text{H NMR}$: δ 2.29 (3H, s), 2.52 (3H, s), 5.03 (1H, $J = 3.3, 5.0$ Hz, H-2), 3.44 (1H, dd) 3.34 (1H, dd) 6.89 (1H, d , $J = 8$ Hz) 7.81 (2H, m); MS m/z : 204 $[\text{M}]^+$, 189, 161, 145, 118, 89, 63, 43 (100%).

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SESQUITERPENE LACTONES FROM *CENTAUREA CORONOPIFOLIA*

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Key Word Index—*Centaurea coronopifolia*; Compositae; sesquiterpene lactones; germacranolides; flavonoids.

Abstract—The aerial parts of *Centaurea coronopifolia* afforded three new germacranolides, two of which are closely related to balsamin. The third is a new derivative of stizolicin, which is also present. The aerial parts also yielded four 6-methoxylated flavonoids and the triterpene, α -amyrin. The structures of the compounds were determined by spectroscopic methods.

INTRODUCTION

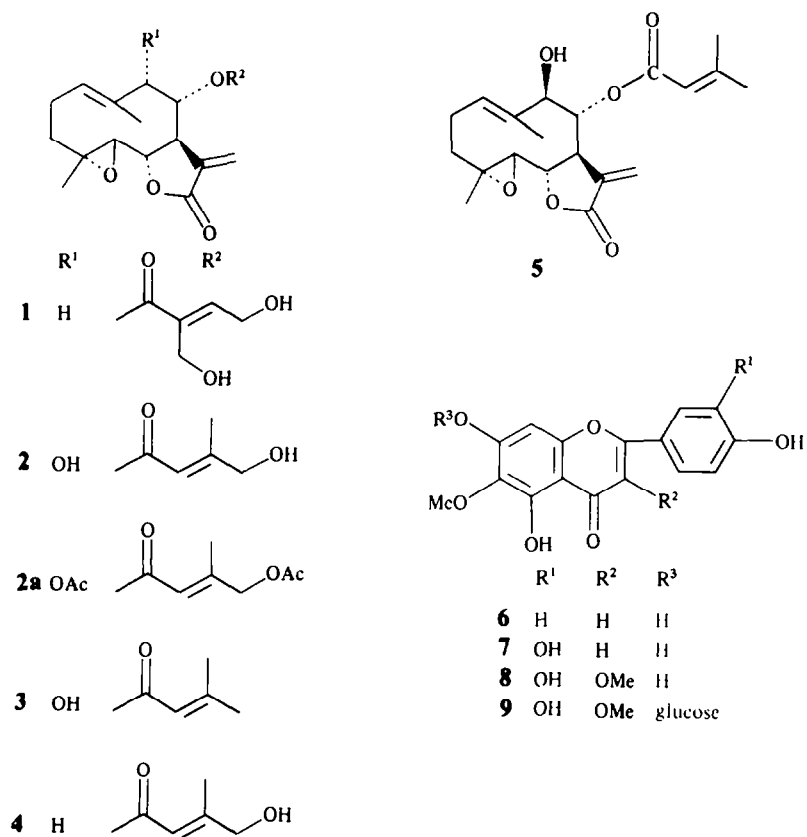
In the course of our investigation on the genus *Centaurea* [1–4], we have isolated four sesquiterpene lactones from the leaves of *Centaurea coronopifolia* Lam. Although one of the lactones, stizolicin, was isolated from *C. coronopifolia* [5] under the syn. *Stizolophus coronopifolius*, the other three are new: 1(10)en-4 α -5 β -epoxy-9 α -hydroxy germacranolides with 8 α -(4-hydroxy senecioate) (2), 8 α -senecioate (3) and 1(10)en-4 α -5 β -epoxy-8 α -(4-hydroxy senecioate) (4) side chains.

In addition, we also isolated the 6-methoxylated flavonoids 6-methoxyapigenin (6), 6-methoxyluteolin (7), quercetagenin 3,6-dimethyl ether (8) and quercetagenin 3,6-dimethyl ether 7-O-glucoside (9). The last two flavonoids were isolated from a *Centaurea* species for the first time. The leaves also yielded large amounts of the well known triterpene, α -amyrin.

RESULTS AND DISCUSSION

Compound 2 had the molecular formula $\text{C}_{20}\text{H}_{26}\text{O}_7$ (EIMS). Its IR spectrum showed the presence of a γ -lactone (1745 and 1140 cm^{-1}), an α,β -unsaturated ester function (1705 and 1280 cm^{-1}) and hydroxyl groups (3400 cm^{-1}). Its structure was established by $^1\text{H NMR}$ spectroscopy.

The $^1\text{H NMR}$ spectrum of 2 (Table 1) contained doublets for an exocyclic methylene conjugated with a γ -lactone at δ 6.29 and 5.76 ($^4J = 3.5, 3$ Hz). A doublet at δ 2.62 ($J = 9$ Hz) for H-5 supported an epoxide group between C-4/C-5. This doublet collapsed to a singlet when the double doublet at δ 4.33 ($J = 6.5, 9$ Hz) (H-6) was irradiated and, in addition, the signal at δ 3.82 (dddd, H-7) was simplified. However, irradiation of H-7 did not affect the double doublet at δ 2.62 (H-5) but converted the double doublet at δ 4.33 to a doublet ($J = 9$ Hz), the

Table 1. ^1H NMR spectral data of compounds 1–4 (CHCl_3 , TMS as int. standard)

H	1*	2†	2a*	3†	4*
1	5.32 <i>m</i>	5.67 <i>m</i>	5.45 <i>dm</i>	5.62 <i>m</i>	5.30 <i>m</i>
2 α	—	1.26 <i>ddd</i>	—	1.36 <i>ddd</i>	—
2 β	—	2.52 <i>dddd</i>	—	2.52 <i>dddd</i>	—
3 α	—	2.22 <i>ddd</i>	—	2.22 <i>ddd</i>	—
3 β	—	2.32 <i>m</i>	—	2.35 <i>m</i>	—
5	2.65 <i>d</i>	2.62 <i>d</i>	2.62 <i>d</i>	2.62 <i>d</i>	2.65 <i>d</i>
6	4.34 <i>dd</i>	4.33 <i>dd</i>	4.40 <i>dd</i>	4.35 <i>dd</i>	4.32 <i>dd</i>
7	3.31 <i>m</i>	3.82 <i>dddd</i>	3.70 <i>dddd</i>	3.81 <i>dddd</i>	3.25 <i>dddd</i>
8	4.58 <i>m</i>	4.57 <i>d</i>	4.55 <i>s (br)</i>	4.57 <i>d</i>	4.60 <i>m</i>
9	—	4.38 <i>s (br)</i>	5.20 <i>s (br)</i>	4.37 <i>s (br)</i>	—
13a	5.70 <i>d</i>	5.67 <i>d</i>	5.56 <i>d</i>	5.66 <i>d</i>	5.75 <i>d</i>
13b	6.32 <i>d</i>	6.29 <i>d</i>	6.30 <i>d</i>	6.29 <i>d</i>	6.33 <i>d</i>
14	1.82 <i>s</i>	1.80 <i>s</i>	1.87 <i>s</i>	1.83 <i>s</i>	1.82 <i>s</i>
15	1.30 <i>s</i>	1.23 <i>s</i>	1.30 <i>s</i>	1.31 <i>s</i>	1.30 <i>s</i>
OCOR	6.99 <i>t</i>	6.02 <i>sm</i>	5.58 <i>q</i>	5.68 <i>q</i>	5.99 <i>q</i>
	4.35 <i>s (br)</i>	4.16 <i>s (br)</i>	4.52 <i>s (br)</i>	1.94 <i>s</i>	4.17 <i>s (br)</i>
	4.80 <i>d</i>	2.02 <i>s</i>	2.07 <i>s</i>	2.13 <i>s</i>	2.03 <i>s</i>
OAc	—	—	2.17 <i>s</i>	—	—
	—	—	2.19 <i>s</i>	—	—

*200 MHz.

†400 MHz.

J (Hz) 2, 2a, 3: 1, 2 β = 7; 2 α , 2 β = 13; 2 β , 3 α = 6; 2 β , 3 β = 13; 3 α , 2 α = 1.5; 3 α , 3 β = 12; 6, 5 = 9; 6, 7 = 6.5; 7, 13a = 3; 7, 13b = 3.5; 7, 8 = 4.

J (Hz) 1 and 4: 5, 6 = 9; 6, 7 = 6; ester $\text{CH}_2\text{-O}$ = 6.

doublets of the exocyclic methylene at $\delta 6.29$ and 5.76 to sharp singlets and the doublet at $\delta 4.57$ ($J = 4$ Hz) to a broadened singlet, indicating that this signal must belong to H-8. The singlet (*br*) at $\delta 4.34$ and the multiplet at $\delta 5.67$ were assigned to H-9 which is adjacent to a hydroxyl group and the vinyl proton H-1 respectively. The protons of the ester side chain were accounted for by a one proton multiplet at $\delta 6.02$, a two proton singlet at $\delta 4.16$ and a three proton singlet at $\delta 2.02$. The chemical shifts of the multiplet and two proton singlet strongly suggested the presence of a 4-hydroxy senecioate moiety. This was confirmed by the presence in the mass spectrum of a base peak at m/z 99.

Acetylation of **2** (to form **2a**) gave rise to two sharp methyl signals at $\delta 2.17$ and 2.19 and caused downfield shifts at $\delta 5.20$ and 4.52 for H-9 and CH_2OH in the side chain, respectively.

The upfield chemical shift of H-8 was explained by the report in the literature [6] that in C-6-*trans*-fused germacranolides with C-8 α -esters, H-8 β has a higher chemical shift than that of H-8 α .

The stereochemistry at C-5 and C-9 followed the observed coupling constants ($J_{5\alpha,6\beta} = 9$ Hz; $J_{8\beta,9\alpha} = 3$ Hz; $J_{8\beta,9\beta} = 0-1$ Hz) when the Dreiding model was inspected.

Compound **3**, $\text{C}_{20}\text{H}_{26}\text{O}_6$ (EIMS) exhibited IR absorption bands at 1740 and 1150 (γ -lactone), 3400 (hydroxyl), and 1708 and 1290 (ester) cm^{-1} . Its ^1H NMR spectrum (Table 1) exhibited the typical doublets of an exocyclic methylene at $\delta 6.29$ and 5.66 ($^4J = 3.5$, 3 Hz). The other signals were very similar to those of compound **2**, except for the side chain signals which consisted of two vinylic methyl singlets at $\delta 1.94$ and 2.13 and a vinylic proton quartet $\delta 5.68$. On the basis of the chemical shift of the vinyl proton, the ester side chain must be senecioate. This was confirmed by the presence in the mass spectrum of a base peak at m/z 83. Compound **3** was identified as the 9α -hydroxy isomer of balsamin (**5**) which was isolated from *Stizolophus balsamita* [7].

Compound **4**, $\text{C}_{20}\text{H}_{26}\text{O}_6$ (EIMS) gave rise to IR bands at 3440 (hydroxyl), 1750 and 1140 (γ -lactone), and 1715 and 1280 (ester) cm^{-1} . Its ^1H NMR spectrum (Table 1) had the characteristic signals of an α -methylene- γ -lactone at $\delta 5.75$ and 6.33 ($^4J = 3$, 3.5 Hz), a H-5 epoxy proton at $\delta 2.65$ (d , $J = 9$ Hz) and a 4-hydroxy senecioic group at $\delta 5.99$ (m , vinylic proton), 2.03 (s , vinyl methyl) and 4.17 (s , CH_2OH). The presence of this ester group was confirmed by the base peak at m/z 99. The spectral data indicated that compound **4** is a new derivative of stizolicin (**1**).

EXPERIMENTAL

Mps uncorr. Plant material was collected from Polatli (near Ankara, Turkey) in July 1980. A voucher specimen (No. 45502) is

deposited in the Herbarium of the Faculty of Pharmacy, Istanbul.

Isolation and identification of compounds 1-4. Air-dried and powdered aerial parts (1 kg) of *C. coronopifolia* was extracted with CHCl_3 in a Soxhlet, after evapn *in vacuo* the extract (21 g) was subjected to CC over silica gel and eluted with CHCl_3 by gradual addition of MeOH.

Compound 1. Mp 141° , white crystals from MeOH (100 mg). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3400 (OH), 1760 and 1150 (γ -lactone), 1715 and 1276 (ester), 1649 ($\text{C}=\text{C}$), $840-800$ ($\text{HC}=\text{CRR}'$). ^1H NMR: see Table 1. MS m/z (rel. int.): 378 [M] $^+$ $\text{C}_{20}\text{H}_{26}\text{O}_7$ (**5**), 360 [$\text{M} - 18$] $^+$ (**10**), 263 [$\text{M} - \text{C}_3\text{H}_7\text{O}_3$] $^+$ (**8**), 246 [$\text{M} - \text{C}_3\text{H}_7\text{O}_4$] $^+$ (**25**), 188 [$\text{C}_{12}\text{H}_{12}\text{O}_2$] $^+$ (**92**), 97 [$\text{C}_3\text{H}_5\text{O}_2$] $^+$ (**100**). UV $\lambda_{\text{max}}^{\text{MeOH}}$: 225 nm.

Compound 2. Mp 203° , white crystals from MeOH (20 mg). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3400 (OH), 1745 and 1140 (γ -lactone), 1705 and 1280 (ester), 1645 ($\text{C}=\text{C}$), $840-800$ ($\text{HC}=\text{CRR}'$). ^1H NMR: see Table 1. MS m/z (rel. int.): 378 [M] $^+$ $\text{C}_{20}\text{H}_{26}\text{O}_7$ (**7**), 279 [$\text{M} - \text{C}_3\text{H}_7\text{O}_2$] $^+$ (**8**), 262 [$\text{M} - \text{C}_3\text{H}_8\text{O}_3$] $^+$ (**24**), 115 [$\text{C}_3\text{H}_7\text{O}_3$] $^+$ (**10**), 99 [$\text{C}_3\text{H}_7\text{O}_2 - \text{CO}$] $^+$ (**68**). UV $\lambda_{\text{max}}^{\text{MeOH}}$: 227 nm. **Diacetate of 2.** A mixture of **2** (5 mg), $\text{C}_2\text{H}_5\text{N}$ (1 ml) and Ac_2O (1 ml) was left at room temp. for 12 hr. ^1H NMR: see Table 1. MS m/z (rel. int.): 462 [M] $^+$ $\text{C}_{24}\text{H}_{30}\text{O}_9$ (**4**), 378 [$\text{M} - \text{C}_4\text{H}_4\text{O}_2$] $^+$ (**6**), 141 [$\text{C}_7\text{H}_9\text{O}_3$] $^+$ (**40**), 99 [$\text{C}_3\text{H}_7\text{O}_2$] $^+$ (**100**).

Compound 3. Mp 225° , white crystals from MeOH (13 mg). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3400 (OH), 1740 and 1150 (γ -lactone), 1708 and 1290 (ester), 1645 ($\text{C}=\text{C}$), $850-818$ ($\text{HC}=\text{CRR}'$). ^1H NMR: see Table 1. MS m/z (rel. int.): 362 [M] $^+$ $\text{C}_{20}\text{H}_{26}\text{O}_6$ (**5**), 279 [$\text{M} - \text{C}_3\text{H}_7\text{O}$] $^+$ (**23**), 263 [$\text{M} - \text{C}_3\text{H}_7\text{O}_2$] $^+$ (**5**), 244 [$\text{C}_{13}\text{H}_{18}\text{O}_4 - \text{H}_2\text{O}$] $^+$ (**12**), 83 [$\text{C}_3\text{H}_7\text{O}$] $^+$ (**100**), 55 [$\text{C}_3\text{H}_7\text{O} - \text{CO}$] $^+$ (**64**). UV $\lambda_{\text{max}}^{\text{MeOH}}$: 230 nm.

Compound 4. Oily (5 mg). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3440 (OH), 1750 and 1140 (γ -lactone), 1715 and 1280 (ester), 1655 ($\text{C}=\text{C}$). ^1H NMR: see Table 1. MS m/z (rel. int.): 362 [M] $^+$ $\text{C}_{20}\text{H}_{26}\text{O}_6$ (**5**), 264 [$\text{M} - \text{C}_3\text{H}_8\text{O}_2$] $^+$ (**5**), 247 [$\text{M} - \text{C}_3\text{H}_8\text{O}_3$] $^+$ (**9**), 99 [$\text{C}_3\text{H}_7\text{O}_2$] $^+$ (**100**), 83 [$\text{C}_3\text{H}_7\text{O}$] $^+$ (**92**). UV $\lambda_{\text{max}}^{\text{MeOH}}$: 225 nm.

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